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ORGANIZING THE ELEMENTS

Section Review

Objectives

- Explain how elements are organized in a periodic table
- Compare early and modern periodic tables
- Identify three broad classes of elements

Vocabulary

- periodic law
- metals
- nonmetals
- metalloids

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

Chemists used the <u>1</u> of elements to sort them into groups.	1
The periodic table organizes the elements into vertical2	
and horizontal 3 in order of increasing 4 . The table is	3
constructed so that elements that have similar chemical properties	4
are in the same <u>5</u> . <u>6</u> have a high luster, or sheen,	5
when cut. Most nonmetals are <u>7</u> at room temperature.	6
Elements with properties that are similar to those of metals	7
and nonmetals are called <u>8</u> . Across the periodic table,	8
the properties of elements become9 metallic and	9
10 nonmetallic.	10

Part B True-False

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

- **10.** In his periodic table, Mendeleev arranged the elements in order of atomic number.
- **11.** There are six periods in a periodic table.
- **12.** Most of the elements in the periodic table are metals.
- **13.** The elements within a period have similar properties.

Part C Matching

Match each description in Column B to the correct term in Column A

	Column A	Column B
14.	metals a.	a vertical column of elements in the periodic table
15.	periods b.	good conductors of heat and electric current
16.	group c.	poor conductors of heat and electric current
17.	nonmetals d.	have properties that are similar to those of metals and nonmetals
18.	metalloids e.	the horizontal rows of the periodic table

Part D Questions and Problems

Answer the following questions in the space provided.

19. List the elements of Group 5A. Tell whether each is a metal, nonmetal, or metalloid.

20. List three properties of metals.

21. Name two elements that have similar properties to those of chlorine.

Date _____

Class _____