

# 3.3

## CONVERSION PROBLEMS

### Section Review

#### Objectives

- Construct conversion factors from equivalent measurements
- Apply the techniques of dimensional analysis to a variety of conversion problems
- Solve problems by breaking the solution into steps
- Convert complex units, using dimensional analysis

#### Vocabulary

- conversion factor
- dimensional analysis

#### Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

Whenever two measurements are equal, or equivalent, **1.** \_\_\_\_\_  
 a ratio of these two measurements will equal **1**. **2.** \_\_\_\_\_  
 A ratio of equivalent measurements is called a **2**. When **3.** \_\_\_\_\_  
 a measurement is multiplied by a conversion factor, the value **4.** \_\_\_\_\_  
 of the measurement **3**. **5.** \_\_\_\_\_  
 In **4**, the units that are a part of the measurements **6.** \_\_\_\_\_  
 are used to help solve the problem. The form of the conversion **7.** \_\_\_\_\_  
 factor that is used is the one in which the unit of the **5** is **8.** \_\_\_\_\_  
 in the denominator. **9.** \_\_\_\_\_  
 Many complex word problems can be solved by breaking the **10.** \_\_\_\_\_  
 solution into **6**. When converting between units, it is often  
 necessary to use more than one **7**.  
 In doing multistep problems, it is important to check that the  
 numerator and **8** of each conversion factor are equivalent.  
 When the **9** cancel, you should be left with the unit of the **10**.

## Part B True-False

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

- \_\_\_\_\_ 11. The units of a conversion factor must cancel.
- \_\_\_\_\_ 12. The conversion factor for changing between grams and milligrams is  $\frac{1 \text{ g}}{1000 \text{ mg}}$ .
- \_\_\_\_\_ 13. Multiple conversion factors can be used to solve complex conversion problems.
- \_\_\_\_\_ 14. If density = mass/volume, then mass = density/volume.
- \_\_\_\_\_ 15. When two measurements are equal, a ratio of these two measurements will equal unity.

## Part C Questions and Problems

Answer the following in the space provided.

16. Make the following conversions using Tables 3.1 and 3.2. Write your answers in scientific notation.
- 125 g to kilograms
  - 0.12 L to mL
17. If 1500 white blood cells are lined up side by side, they would form a row 1.0 inch long. What is the average diameter in micrometers of a single white blood cell? (1 inch = 2.54 cm)
18. A radio wave travels 186,000 miles per second. How many kilometers will the wave travel in one microsecond? (1 mile = 1.61 km)