MIXTURES

Section Review

Objectives

- Classify a sample of matter as a substance or a mixture
- Distinguish between homogeneous and heterogeneous samples of matter
- Describe two ways that components of mixtures can be separated

Vocabulary

- mixture
- heterogeneous mixture
- homogeneous mixture
- solution

- phase
- filtration
- distillation

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

A physical blend of two or more substances is a1	1
A mixture has a composition that varies. Mixtures may be identified	2
as2 or3 Homogeneous mixtures are also known	3
as4 and have uniform properties. Any part of a sample	4
with uniform composition and properties is called a5	5
Many mixtures can be separated into their components by	6
6 methods7 is a method of separation that involves	7
boiling a liquid, which is then condensed.	

Part B True-False

_____11. Solutions are liquids.

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

8 .	Homogeneous mixtures can be separated by distillation.
9.	A solution has a uniform composition.
10.	A heterogeneous mixture contains two or more phases.

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Part C Matching

Match each description in Column B to the correct term in Column A.

Column A

Column B

12. mixture

_____ 13. heterogeneous mixture

b. any part of a sample that has uniform composition and

a. a mixture that has a uniform composition throughout

properties

14. homogeneous mixture

_____ **15.** solution

d. separation of a liquid by boiling followed by condensation

_____ **16.** phase

_____ **17.** distillation

_____ 18. filtration

e. another name for a homogeneous mixture

f. a physical blend of two or more components

c. a mixture that is not uniform in composition

g. a method for separating a solid from a liquid in a heterogeneous mixture

Part D Questions and Problems

Answer each of the following questions in the space provided.

19. State whether each of the following is a homogeneous or heterogeneous mixture.

a. table salt dissolved in water

b. carbon mixed with sand

c. filtered apple juice

d. vegetable soup

e. fresh squeezed lemonade

20. Classify each of the following as a substance or a mixture.

a. table sugar (sucrose)

b. hot tea

c. table salt (sodium chloride)

d. vinegar